

Borrowing Hydrogen Catalysis (Hydrogen Autotransfer)

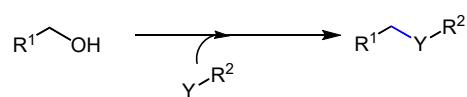
Literature Review

May Group

5/22/18

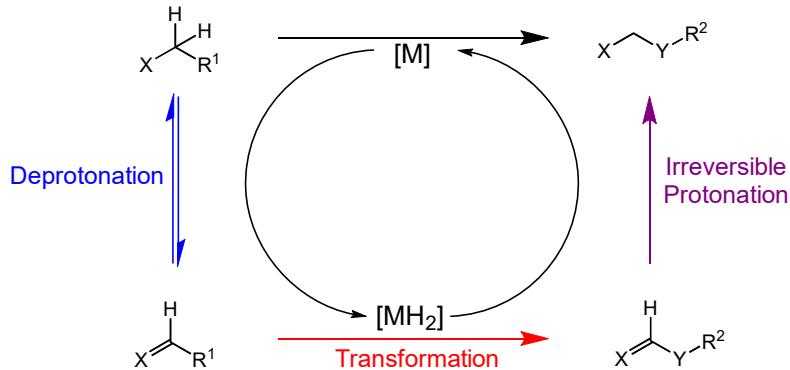
Sasha Oleynichenko

Borrowing Hydrogen (BH) Overview



- First reported in 1908
Hebd. Seances Acad. Sci. **1908**, 146, 298.
- Much progress in the last 20 years
- Direct combination of transfer hydrogenation with an intermediate reaction
- Can form complex molecules without separation/isolation

Borrowing Hydrogen (BH) Overview

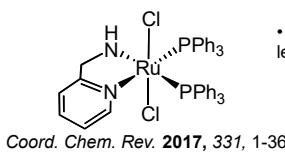


Chem. Rev. 2018, 118, 1410-1459

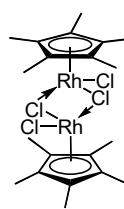
Catalyst Identity

Any metal used must form a sufficiently unstable hydride - walk the line

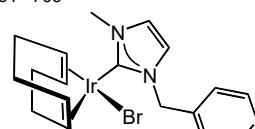
- Most common catalysts are $[\text{Ru}]$, $[\text{Rh}]$, and $[\text{Ir}]$
- $[\text{Os}]$, $[\text{Mo}]$ and $[\text{W}]$ have been reported but are less common



Coord. Chem. Rev. 2017, 331, 1-36

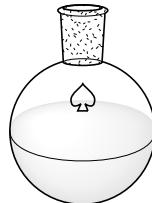


Angew. Chem. Int. Ed. 2014, 53, 761–765
(Hot paper)



Chem. - Eur. J. 2015, 21, 17877-17889

Homogenous vs Heterogenous Catalysis


Homogenous
Advantages:

Lower temperatures
Potential for more active catalyst systems
Great deal of research in the field

Disadvantages:

Generally requires additives (ligands)
Less atom economical


Heterogenous
Advantages:

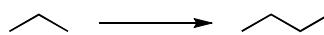
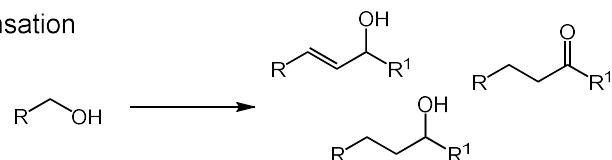
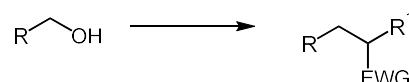
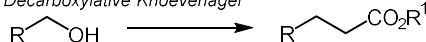
More atom economical (additive free)
Can use multi-functional solid catalysts
Green chemistry appealing to industry

Disadvantages:

High temperatures
Origin of BH methodology, but much less developed

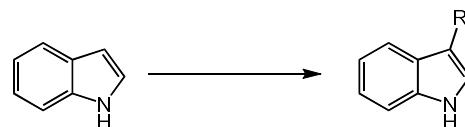
Chem. Rev. 2018, 118, 1410-1459

C-C Bond Formation

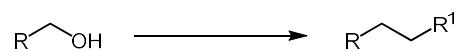
Olefin Metathesis

Aldol Condensation

Knoevenagel Reaction

Decarboxylative Knoevenagel


C-C Bond Formation

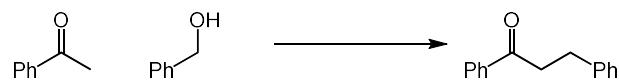
C₃ Alkylation



Wittig Reaction

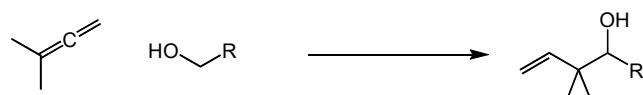


Condensation



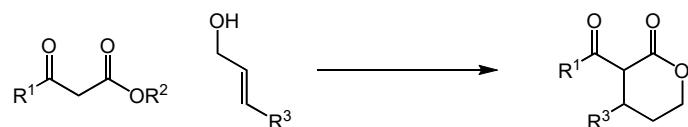
C-C Bond Formation

Allylation



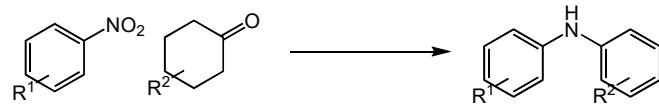
Krische JACS 2007

δ -Lactonization

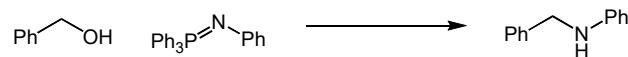


C-N Bond Formation

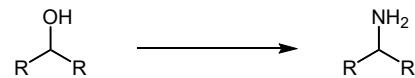
Condensation



Aza-Wittig

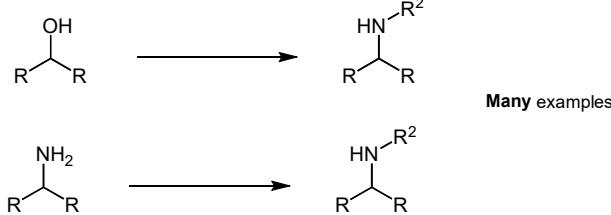


Amination

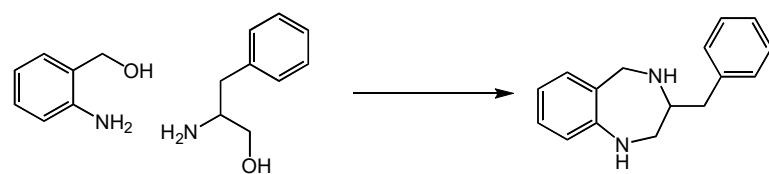


C-N Bond Formation

N-Alkylation

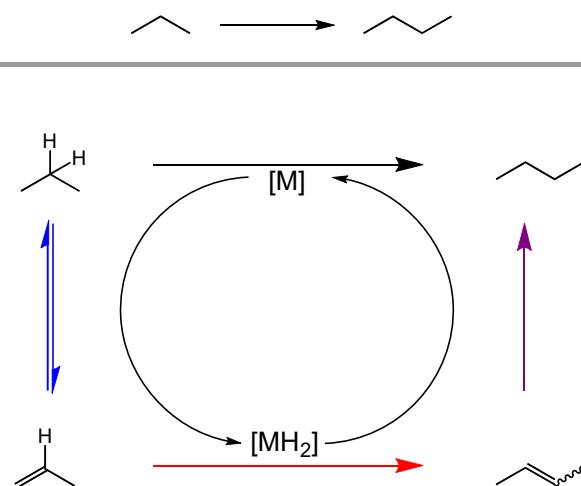


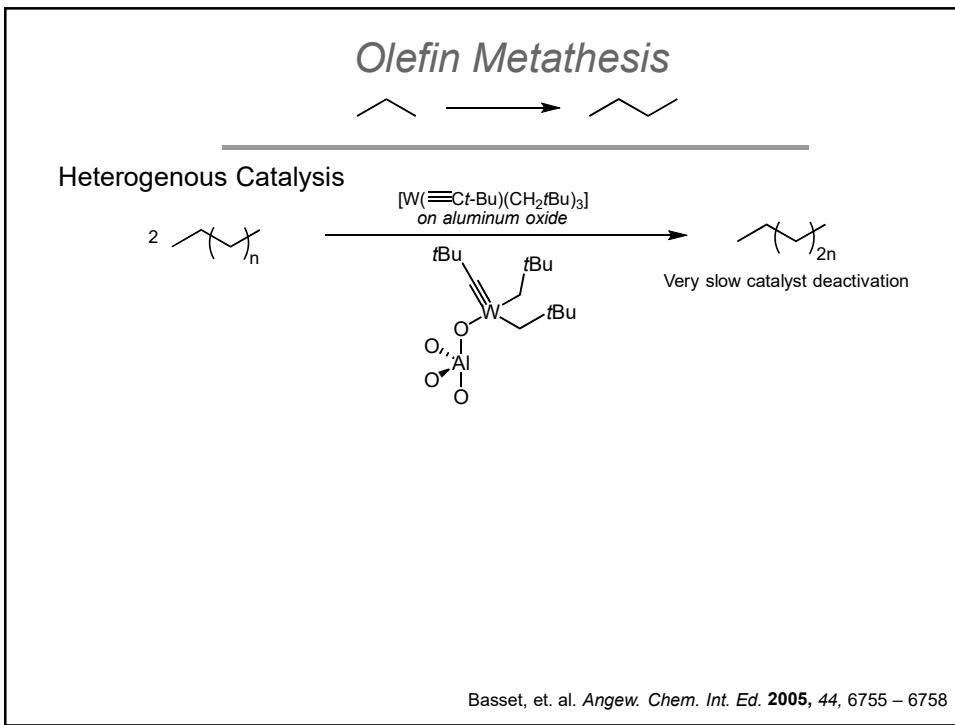
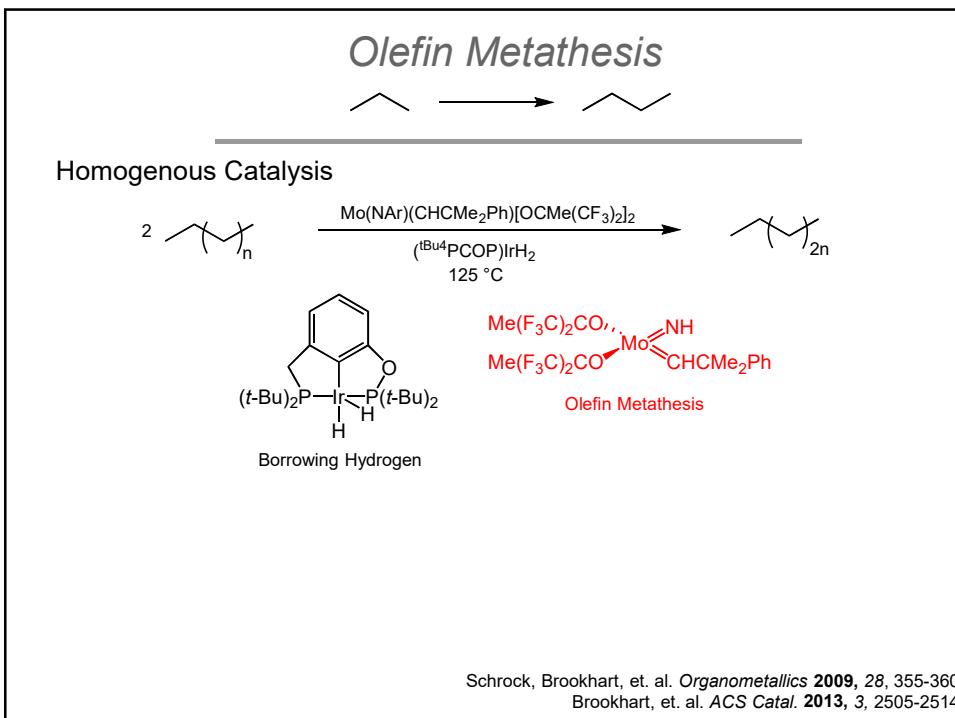
Condensation/N-Heterocyclization

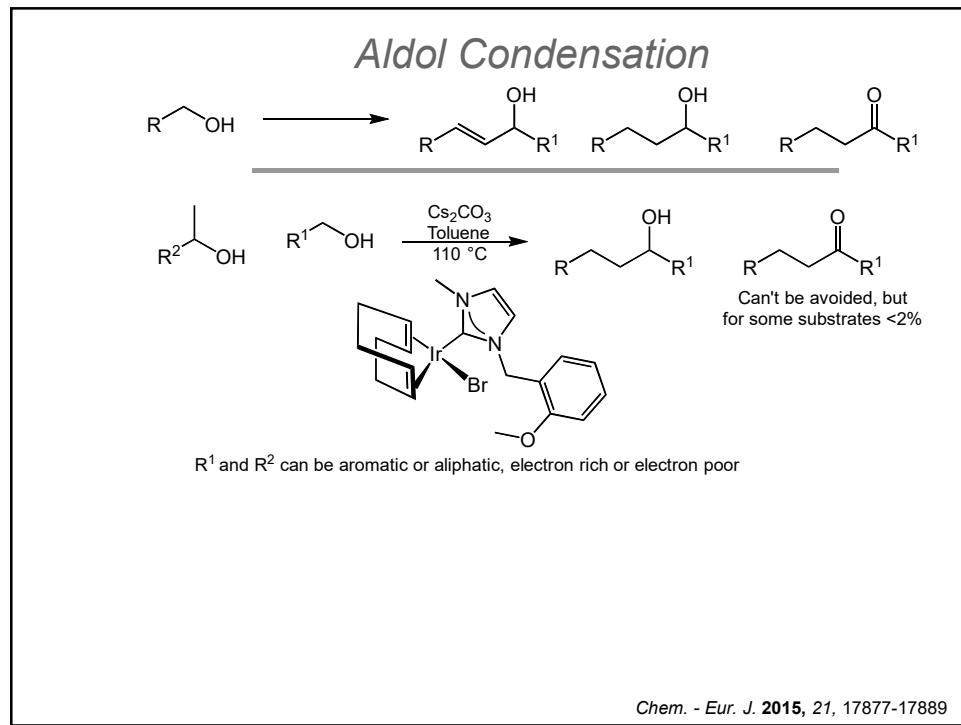
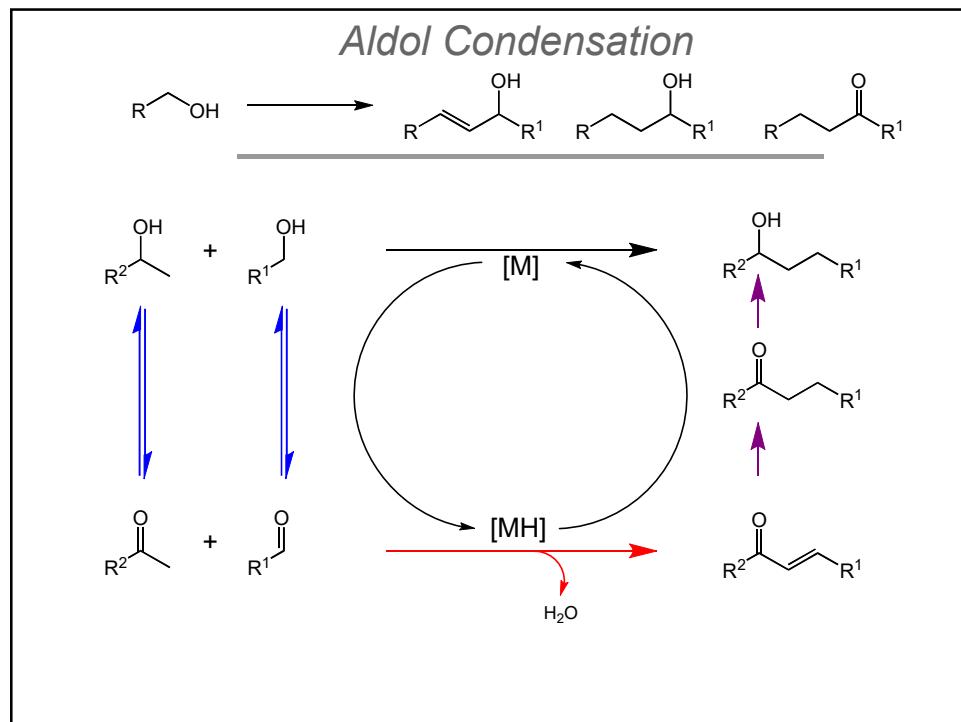


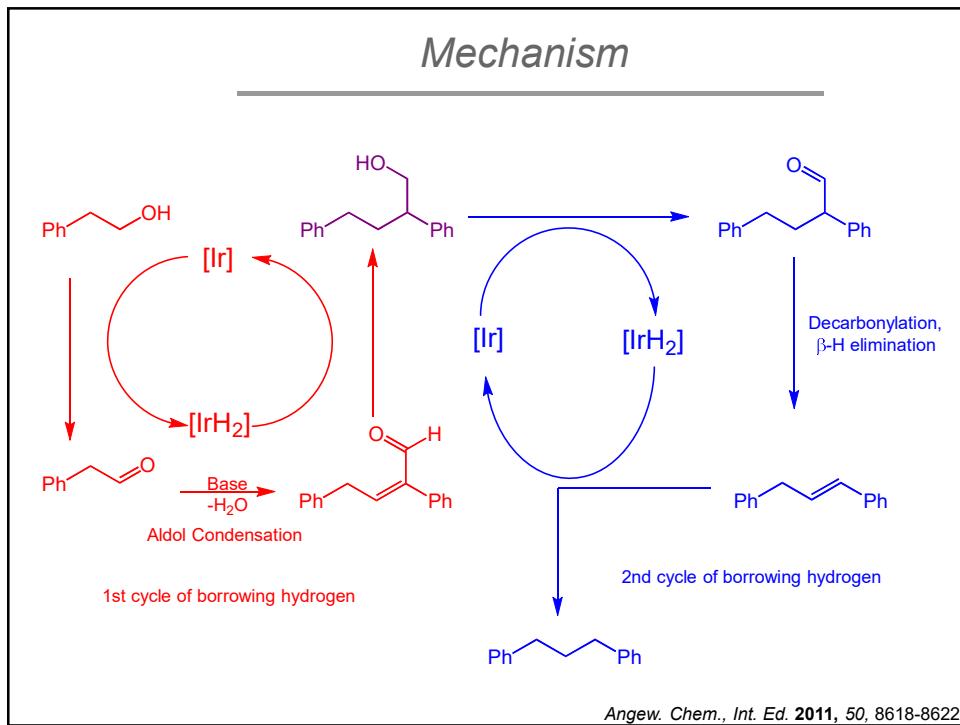
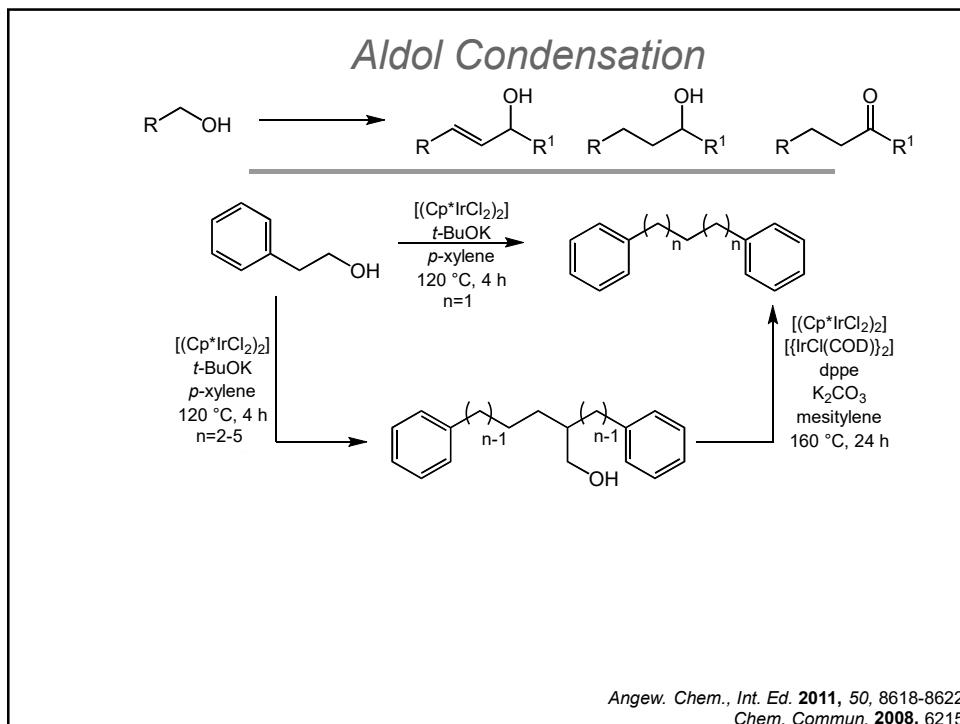
Carbon-Carbon Bond Formation

Olefin Metathesis

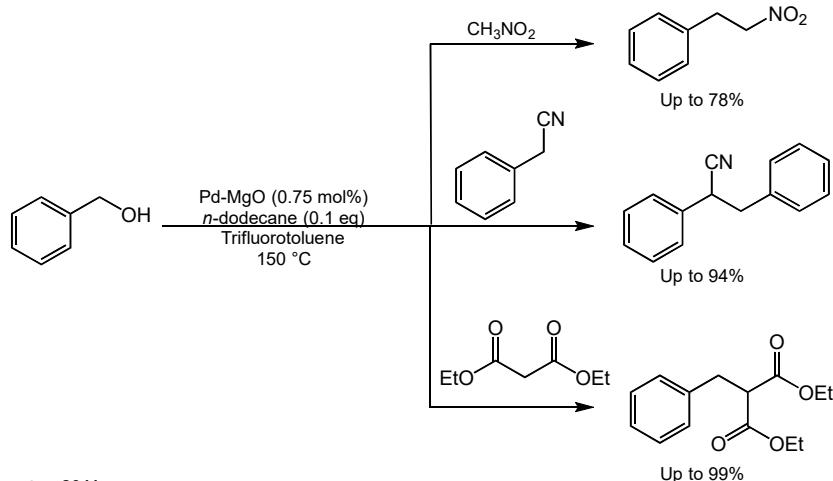
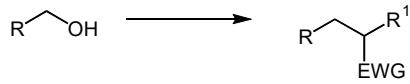




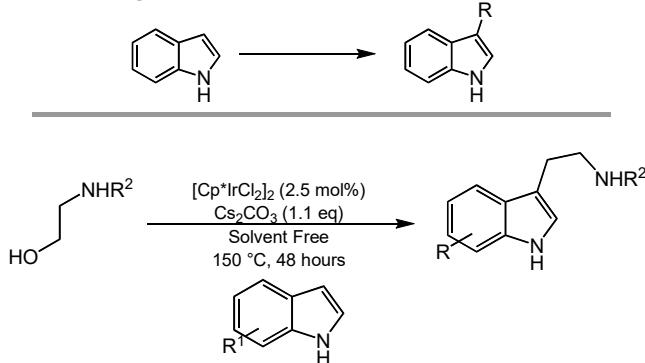




Knoevenagel Condensation

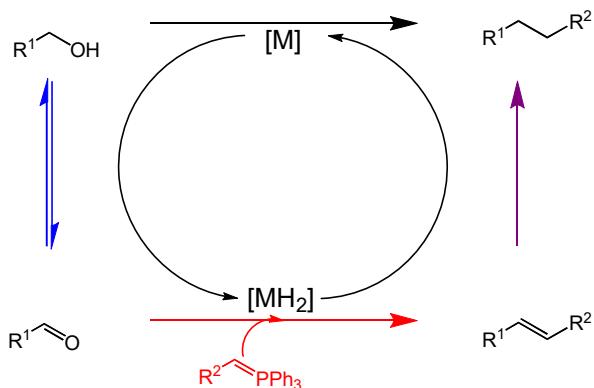
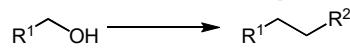
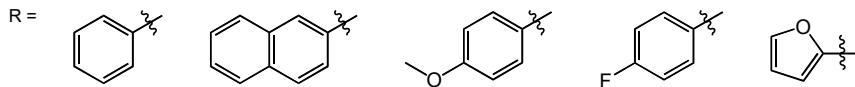
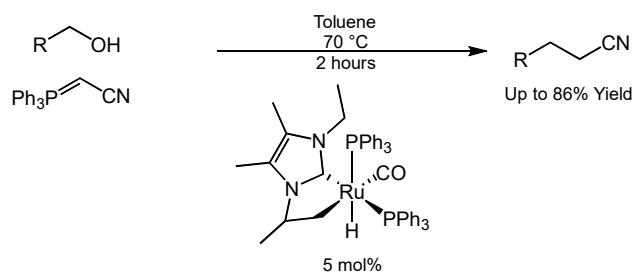
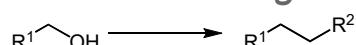


C₃ Activation of Indoles

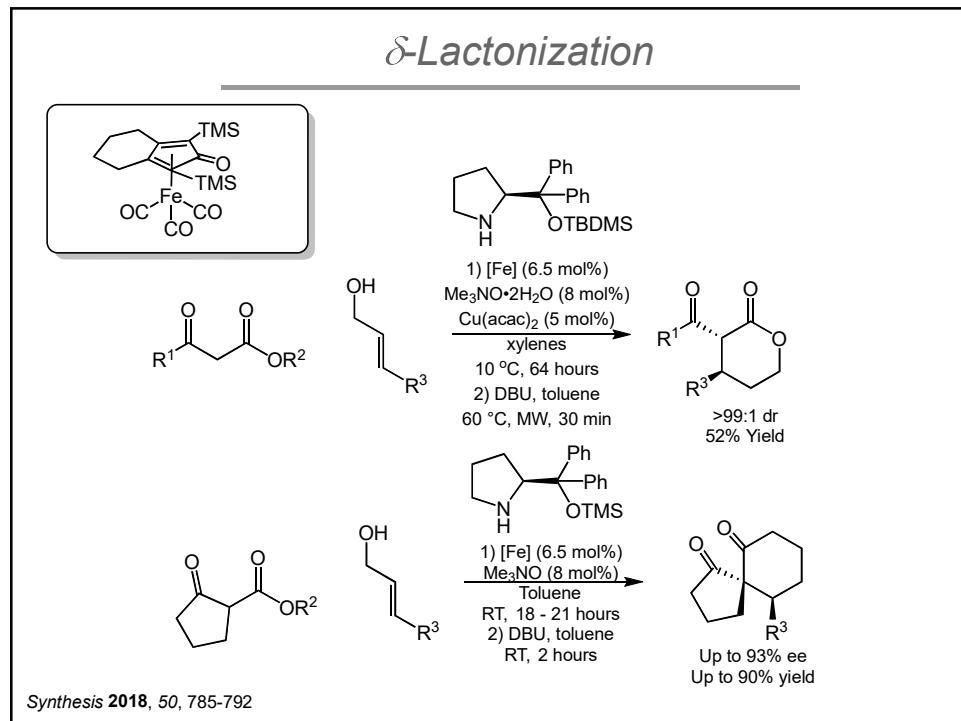
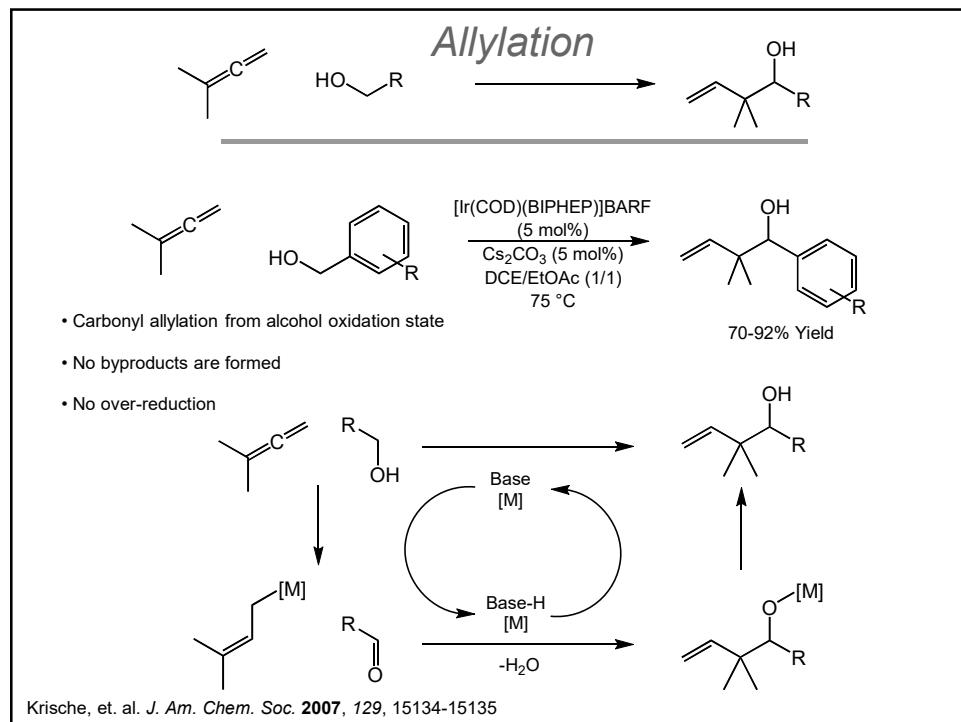


- After deprotection, gives access to many tryptamine homologues, including serotonin
- Branched alcohols are just as reactive
- Longer chains equally reactive

Tetrahedron 2016, 72, 2233-2238

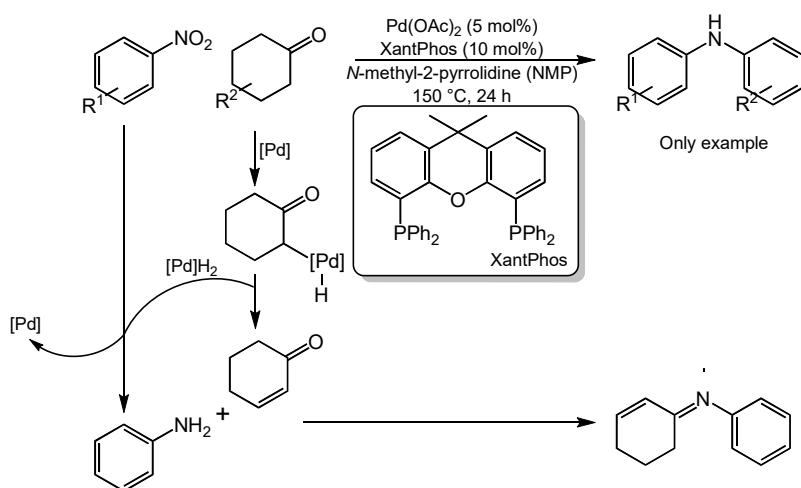
Indirect Wittig*Indirect Wittig*

J. Am. Chem. Soc. **2007**, *129*, 1987-1995

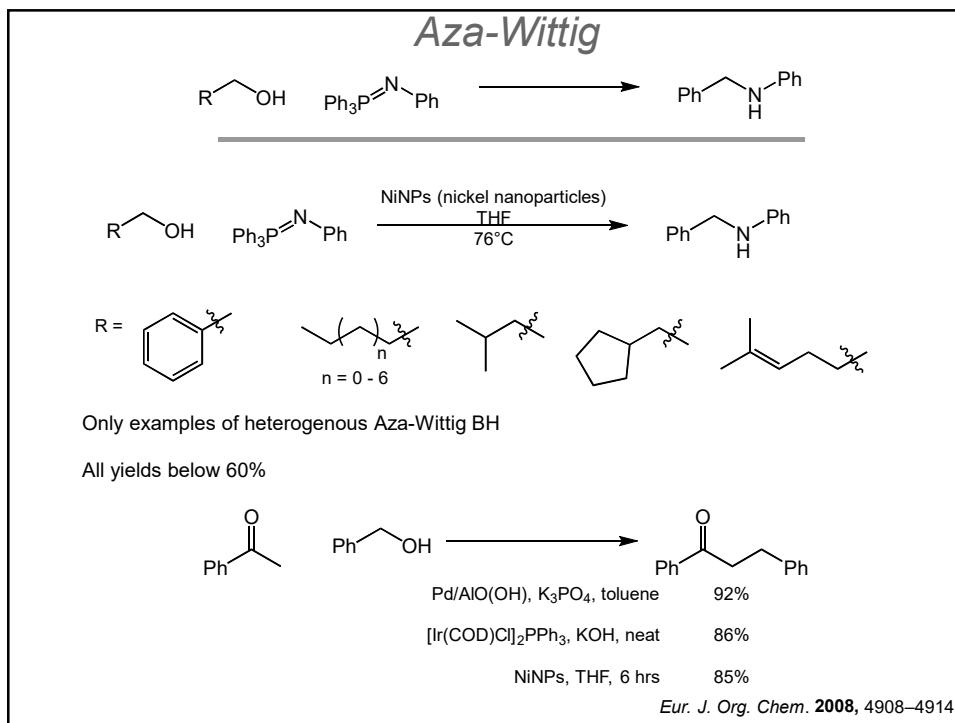
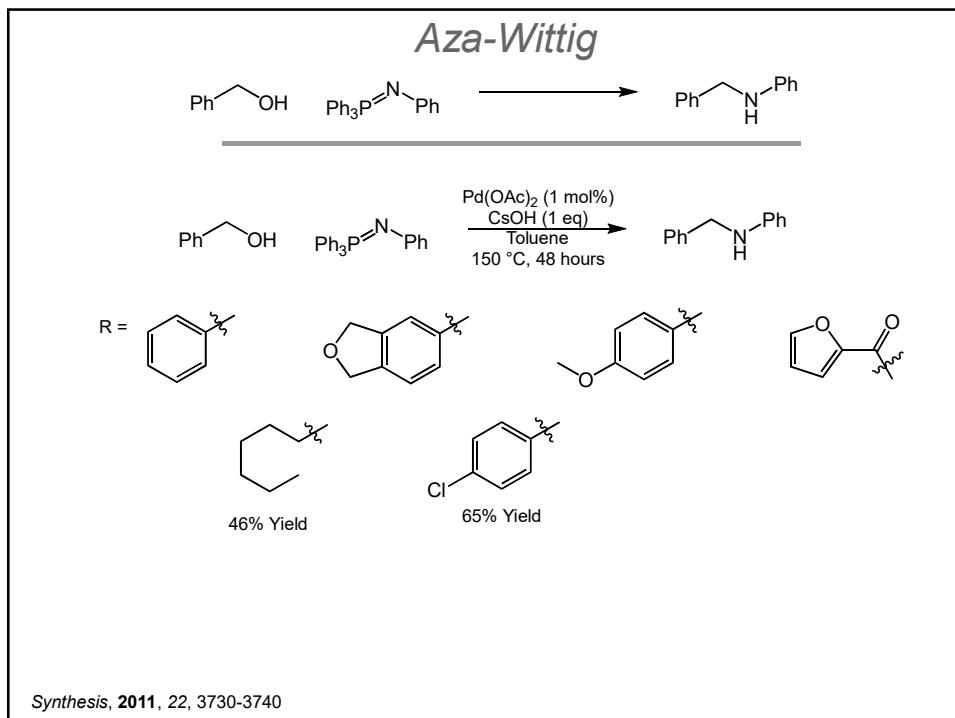


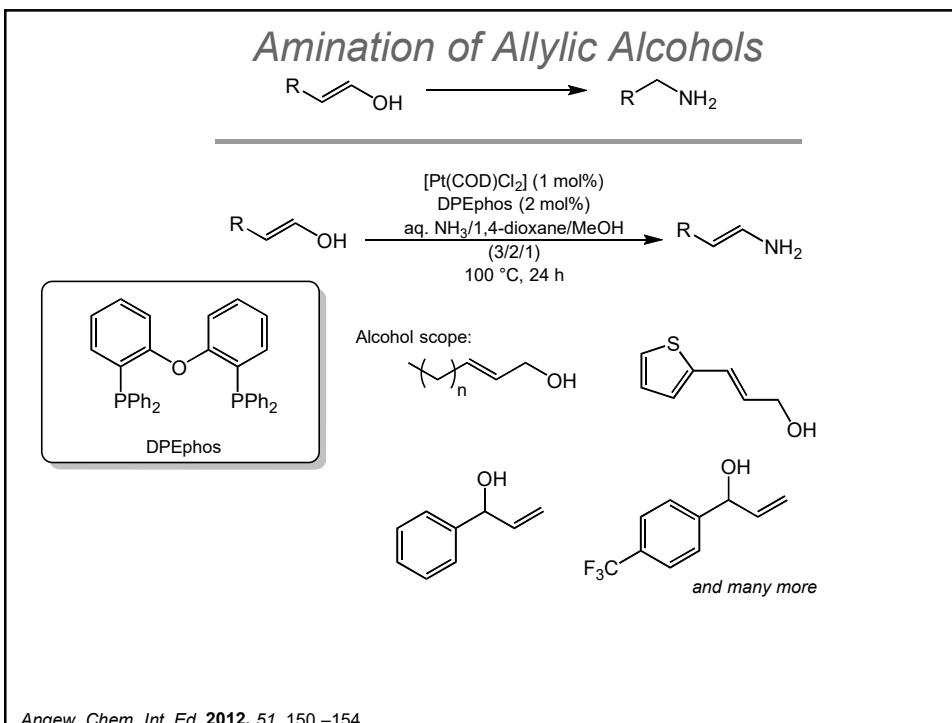
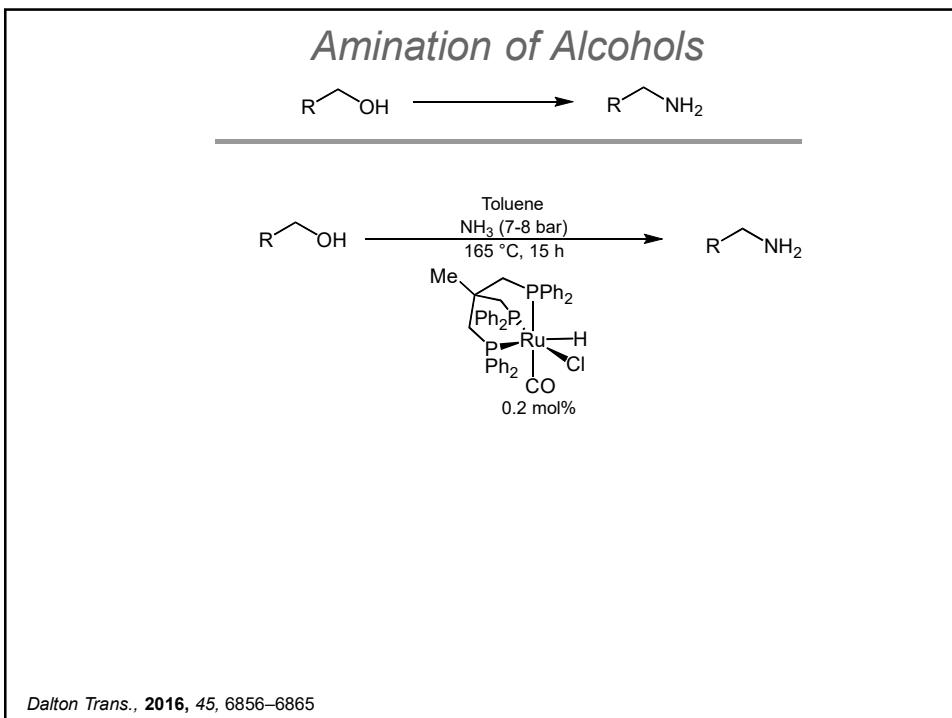
Carbon-Nitrogen Bond Formation

Condensation/ N-Arylation

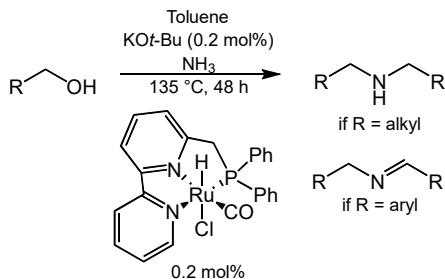
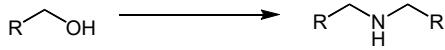


Org. Lett. 2012, 14, 1692-1695



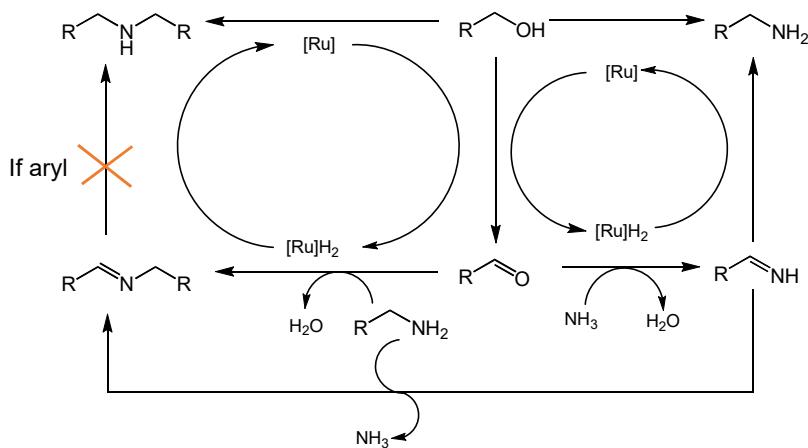


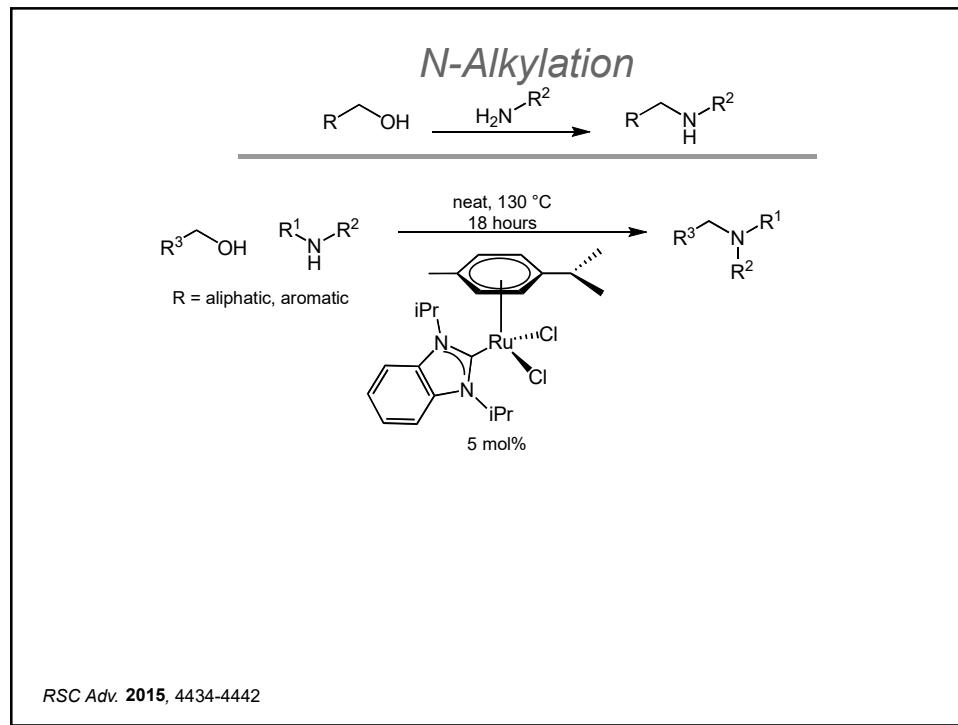
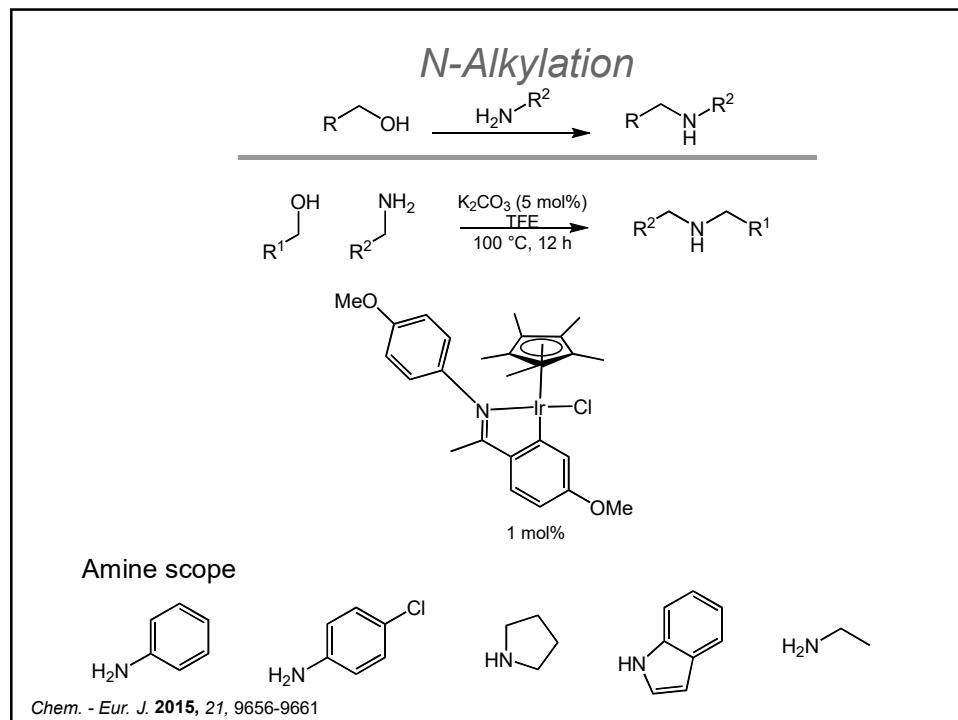
Amination to 2° Amines

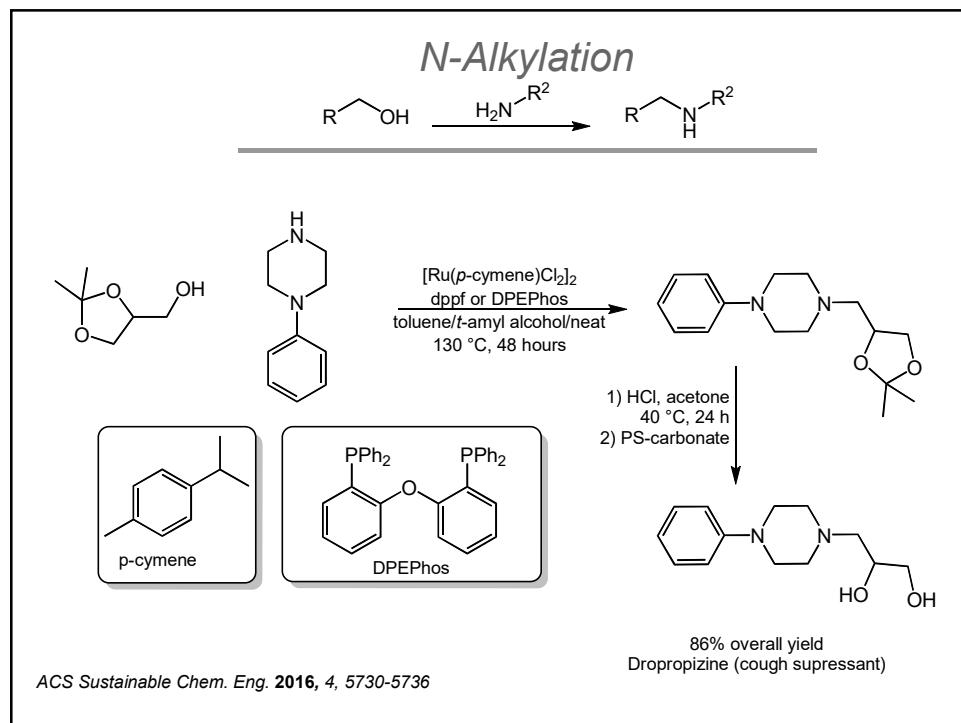
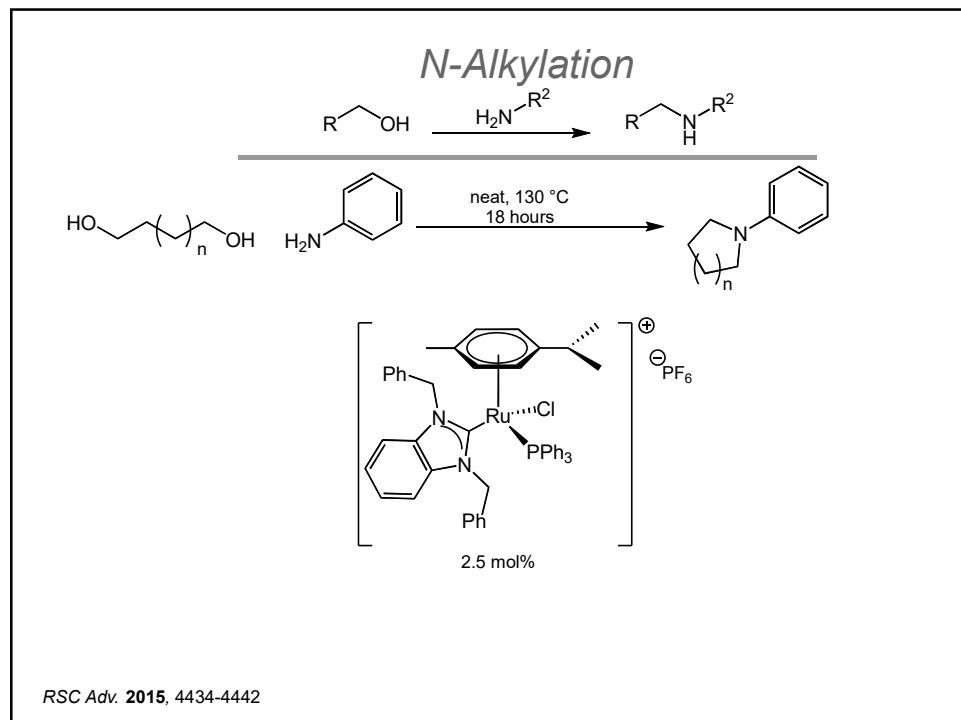


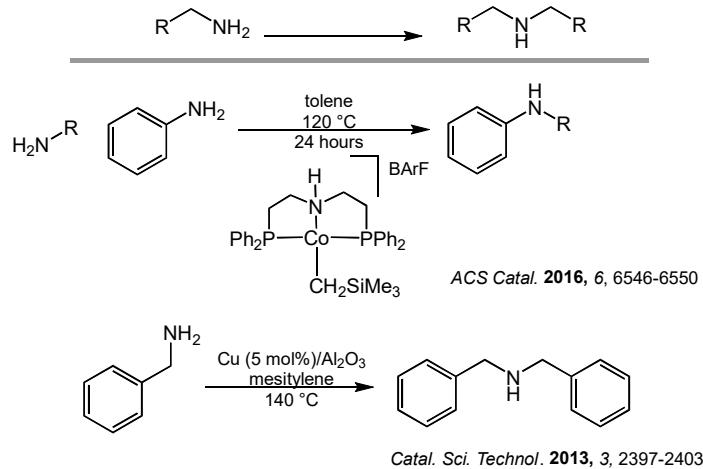
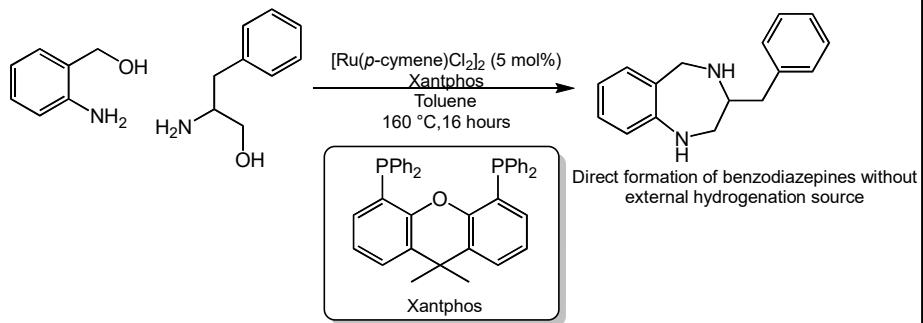
Catal. Lett. 2015, 145, 139-144

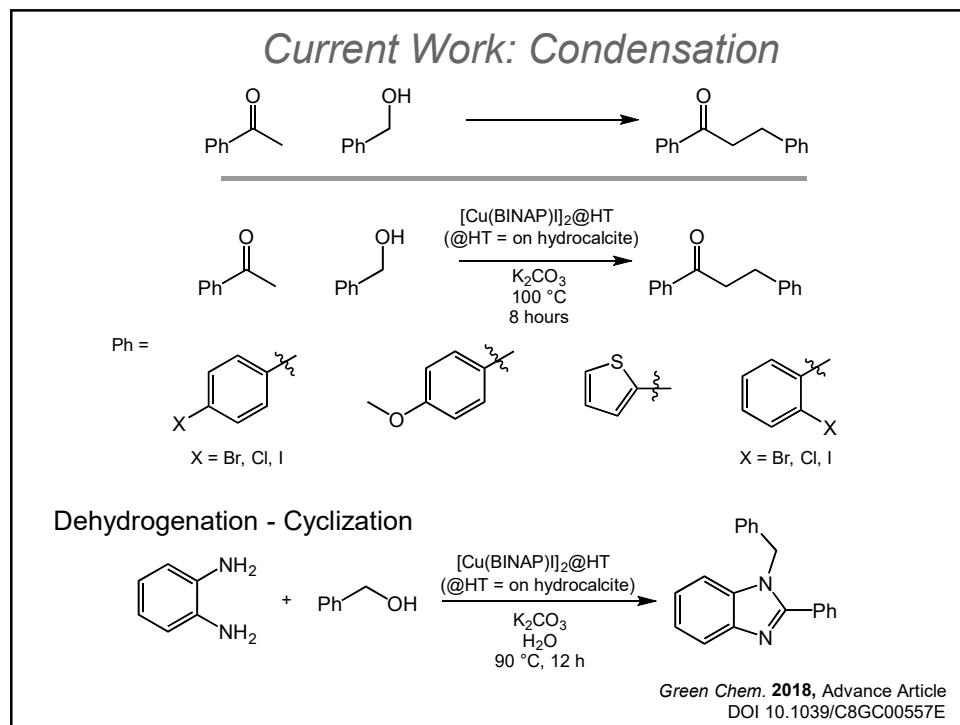
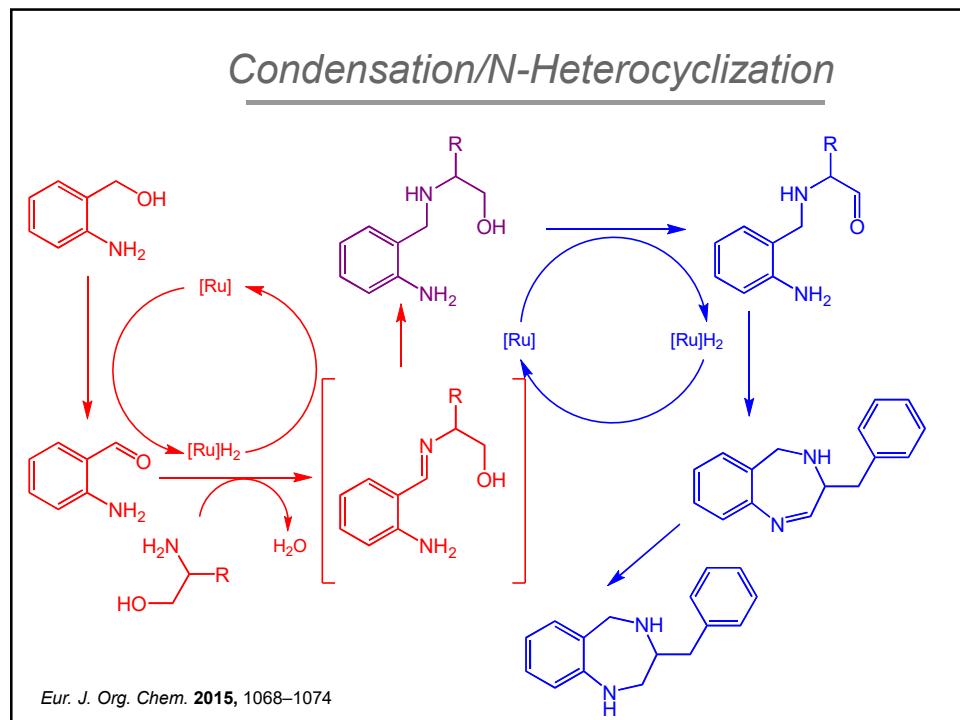
Mechanism



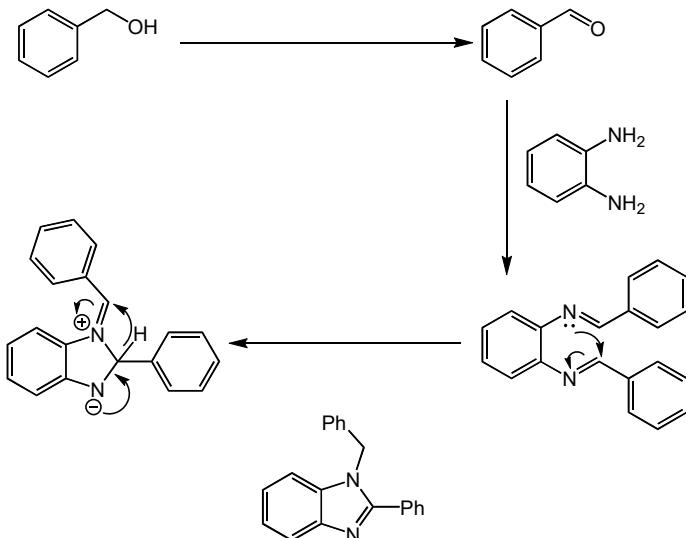




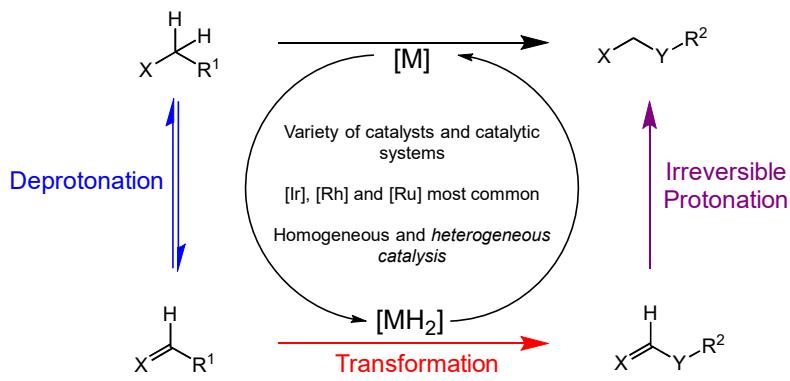
N-Alkylation*Condensation/N-Heterocyclization**Eur. J. Org. Chem.* **2015**, 1068–1074



Mechanism



Borrowing Hydrogen (BH) Summary



C-C bond formation:

Olefin metathesis, **Aldol and Knoevenagel condensation**, Wittig reaction, indole C₃ alkylation, allylation, δ -lactonization

C-N bond formation:

Condensation, Aza Wittig reaction, amination, **N-alkylation**, condensation/N-heterocyclization